

**Q.1 For questions (I to VII), circle the correct answer: (1 pt each question)**

(I) An electron is in a one-dimensional trap with zero potential energy in the interior and infinite potential energy at the walls. The ratio  $E_3/E_1$  of the energy for  $n = 3$  to that for  $n = 1$  is:

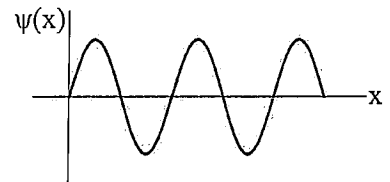
- a. 1/4                      b. 1/9                      c. 3/1                      **d. 9/1**                      e. 4/1

(II) The energy of the third excited state of "an electron in a one-dimensional trap with zero potential energy in the interior and infinite potential energy at the walls" is 32.0 eV. The ground state energy will be:

- a. 4.0 eV                      b. 8.0 eV                      **c. 2.0 eV**                      d. 3.0 eV                      e. None of them

(III) An electron is in a one-dimensional trap with zero potential energy in the interior and infinite potential energy at the walls. A graph of its wave function  $\psi(x)$  versus  $x$  is shown. The value of the quantum number  $n$  is:

- a. 4                      b. 2                      c. 8                      d. 6                      **e. 5**



(IV) If a wave function  $\psi$  for a particle moving along the  $x$  axis is normalized, then:

- a.  $\int |\psi|^2 dt = 1$                       **b.  $\int |\psi|^2 dx = 1$**                       c.  $\partial\psi/\partial x = 1$                       d.  $\partial\psi/\partial t = 1$                       e.  $|\psi|^2 = 1$

(V) For the hydrogen atom an electron is in the state (with the orbital quantum number,  $l = 3$ ), the number of different allowed values for the magnetic quantum number  $m_l$  is equal to:

- a. 7**                      b. 6                      c. 3                      d. 4                      e. none of them

(VI) When a beam of electrons of kinetic energy 40 keV strike a molybdenum target, they produce both continuous and characteristic x-ray. If we increase the kinetic energy of the electrons to 80 keV, then the energy of  $K_\alpha$  line will:

- a. Be doubled                      b. Be half the original value                      **c. not change**  
 d. Be four times the original value                      e. Be decreased to one fourth the original value

(VII) In atomic states a metastable state is a state with:

- a. Life time much smaller than other states                      **b. Life time much larger than other states**  
 c. Life time about that of other states                      d. none of them

**Q.1 For questions (I to VII), circle the correct answer: (1 pt each question)**

(I) An electron is in a one-dimensional trap with zero potential energy in the interior and infinite potential energy at the walls. The ratio  $E_1/E_3$  of the energy for  $n = 1$  to that for  $n = 3$  is:

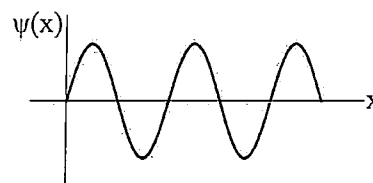
- a. 1/4                       b. 1/9                      c. 3/1                      d. 9/1                      e. 4/1

(II) The energy of the third excited state of "an electron in a one-dimensional trap with zero potential energy in the interior and infinite potential energy at the walls" is 64.0 eV. The ground state energy will be:

- a. 4.0 eV                      b. 8.0 eV                      c. 2.0 eV                      d. 3.0 eV                      e. None of them

(III) An electron is in a one-dimensional trap with zero potential energy in the interior and infinite potential energy at the walls. A graph of its wave function  $\psi(x)$  versus  $x$  is shown. The value of the quantum number  $n$  is:

- a. 4                      b. 2                       c. 5                      d. 6                      e. 8



(IV) If a wave function  $\psi$  for a particle moving along the  $x$  axis is normalized, then:

- a.  $\int |\psi|^2 dt = 1$                       b.  $\partial\psi/\partial t = 1$                       c.  $\partial\psi/\partial x = 1$                        d.  $\int |\psi|^2 dx = 1$                       e.  $|\psi|^2 = 1$

(V) For the hydrogen atom an electron is in the state (with the orbital quantum number,  $l = 4$ ), the number of different allowed values for the magnetic quantum number  $m_l$  is equal to:

- a. 5                      b. 8                       c. 9                      d. 4                      e. none of them

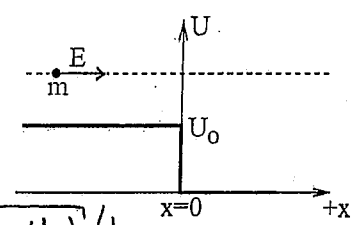
(VI) When a beam of electrons of kinetic energy 40 keV strike a molybdenum target, they produce both continuous and characteristic x-ray. If we increase the kinetic energy of the electrons to 80 keV, then the energy of  $K_\alpha$  line will:

- a. Be doubled                      b. Be half the original value                      c. Be four times the original value  
d. Be decreased to one fourth the original value                       e. not change

(VII) In atomic states a metastable state is a state with:

- a. Life time much larger than other states                      b. Life time much smaller than other states  
c. Life time about that of other states                      d. none of them

**Q.2** An electron of energy  $E = 8 \text{ eV}$  originally traveling to the right in a region of space where  $U_0 = 6 \text{ eV}$  (for  $x < 0$ ) strike "a step down potential of  $U=0$  (for  $x > 0$ )" at  $x=0$ , as shown in the figure. (Use the notations  $k_1$  and  $k_2$  for the wave numbers in the regions  $x < 0$  and  $x > 0$ , respectively).



(i) Write down the general solution to the time-independent Schrödinger equation for the regions  $x < 0$  and  $x > 0$ ? (2 pts)

For  $x < 0$ :  $\psi_1 = A e^{i k_1 x} + B e^{-i k_1 x}$ ,  $k_1 = \sqrt{2m(E - U_0)} / \hbar$   
 For  $x > 0$ :  $\psi_2 = C e^{i k_2 x}$ ,  $k_2 = \sqrt{2mE} / \hbar$

(ii) Use the boundary conditions on the wave function to write linear equations for the unknown constants of your general solution of part (i)? (2 pts)

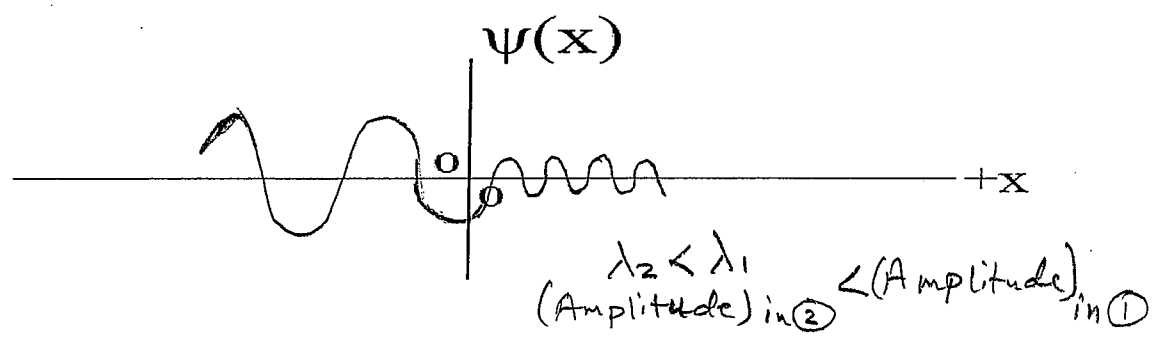
$\psi_1|_{x=0} = \psi_2|_{x=0} \Rightarrow A + B = C$  --- ①  
 $\psi_1'|_{x=0} = \psi_2'|_{x=0} \Rightarrow (A - B)k_1 = C k_2$  --- ②

① \*  $k_1$  + ②  $\Rightarrow 2A k_1 = C (k_1 + k_2) \Rightarrow \frac{C}{A} = \frac{2k_1}{k_1 + k_2}$   
 ① \*  $k_1$  - ②  $\Rightarrow +2B k_1 = C (k_1 - k_2) = \frac{2A k_1 (k_1 - k_2)}{(k_1 + k_2)}$   
 $\frac{B}{A} = \frac{k_1 - k_2}{k_1 + k_2}$

(iii) Find the values for the wave numbers in the regions  $x < 0$  and  $x > 0$ ? (1 pts)

$k_1 = \sqrt{2m_e (E - U_0)} / \hbar = \sqrt{2 \times 9.11 \times 10^{-31} \times (8 - 6) \times 1.6 \times 10^{-19}} / \left( \frac{6.63 \times 10^{-34}}{2\pi} \right)$   
 $\cong 7.24 \times 10^9 \text{ (rad/m)}$   
 $k_2 = \sqrt{2mE} / \hbar = \sqrt{2 \times 9.11 \times 10^{-31} \times 8 \times 1.6 \times 10^{-19}} / \left( \frac{6.63 \times 10^{-34}}{2\pi} \right) \cong 14.47 \times 10^9 \text{ (rad/m)}$

(iv) Sketch the solution, using the x-axis in the following figure. (2 pts)



(v) The reflection coefficient is  $R = [(k_2 - k_1) / (k_2 + k_1)]^2$ , if there are 1000-electrons incident, find the number of transmitted electrons? (2 pts)

$T = 1 - R = 1 - \left[ \frac{k_2 - k_1}{k_2 + k_1} \right]^2 = 1 - \left[ \frac{14.47 - 7.24}{14.47 + 7.24} \right]^2 \cong 0.889$

$T \cong 88.9$   
 Number of transmitted electrons =  $(0.889) * (1000) = 889 \text{ (electrons)}$

Q.3 An electron in the hydrogen atom is in the state  $\psi_{1,0,0} = A e^{-r/a}$

(i) Normalize this wave function (i.e find the value for A that will make the wave function normalized) (2 pts)

$$\int \psi \psi^* dV = 1, \text{ since } \psi \text{ depends on } r \text{ only, then } dV = 4\pi r^2 dr$$

$$\int_0^\infty A^2 e^{-2r/a} 4\pi r^2 dr = 1 \rightarrow 4\pi A^2 \int_0^\infty r^2 e^{-2r/a} dr = 1$$

$$4\pi A^2 \left[ \frac{2!}{(2/a)^3} \right] = \pi A^2 a^3 = 1 \rightarrow A = \frac{1}{\sqrt{\pi} a^{3/2}}$$

(ii) Find the probability that the electron in this state will be found between two spherical shells with radii 0.8a and 1.2a? (2 pts)

$$P = \int_{r=0.8a}^{1.2a} \psi^2 4\pi r^2 dr = \frac{4\pi}{\pi a^3} \int_{0.8a}^{1.2a} r^2 e^{-2r/a} dr = \frac{4}{a^3} \left[ -\frac{1}{(2/a)^3} \left( \left(\frac{2}{a}\right)^2 r^2 + 2\frac{2}{a} r + 2 \right) e^{-2r/a} \right]_{0.8a}^{1.2a}$$

$$= -\frac{1}{2} \left[ \frac{4}{a^2} r^2 + \frac{4}{a} r + 2 \right] e^{-2r/a} \Big|_{0.8a}^{1.2a} = -2 \left[ \frac{r^2}{a^2} + \frac{r}{a} + 0.5 \right] e^{-2r/a} \Big|_{0.8a}^{1.2a}$$

$$= 2 \left[ -\frac{(1.2)^2 a^2}{a^2} + \frac{1.2a}{a} + 0.5 \right] e^{-2(1.2a)/a} + \left[ \frac{(0.8)^2 a^2}{a^2} + \frac{0.8a}{a} + 0.5 \right] e^{-2(0.8a)/a}$$

$$= 2 \left[ -0.28122 + 0.3917 \right] \approx 0.214 \approx 21.4\%$$

Q.4 A rectangular corral of widths  $L_x = L$  and  $L_y = L/2$  contains seven non interacting electrons with spin of  $s = 1/2$  for each electron.

(i) Write down the general form for the energy of any level. (2 pts)

$$E_{n_x, n_y} = n_x^2 \frac{h^2}{8mL_x^2} + n_y^2 \frac{h^2}{8mL_y^2} = n_x^2 \frac{h^2}{8mL^2} + n_y^2 \frac{h^2}{8mL^2/4}$$

$$= (n_x^2 + 4n_y^2) \frac{h^2}{8mL^2}, \quad n_x = 1, 2, 3, \dots$$

$$n_y = 1, 2, 3, \dots$$

(ii) Find the energy of the ground state of this system of seven electrons? (3 pts)

$$E_{1,1} = 5 \frac{h^2}{8mL^2}$$

$$E_{2,1} = 8 \frac{h^2}{8mL^2}$$

$$E_{3,1} = 13 \frac{h^2}{8mL^2}$$

$$E_{1,2} = 17 \frac{h^2}{8mL^2}$$

$$E_{\text{ground}} = 2E_{1,1} + 2E_{2,1} + 2E_{3,1} + 1E_{1,2}$$

$$= [2(5) + 2(8) + 2(13) + 1(17)] \frac{h^2}{8mL^2}$$

$$= 69 \left[ \frac{h^2}{8mL^2} \right]$$

